

A Novel Approach to Bicyclic Alkaloids Using a Tandem Diastereoselective Acyliminocyclization and Retro Diels-Alder Reaction Sequence. Synthesis of (+)-Indolizidine and (+)-Laburnine

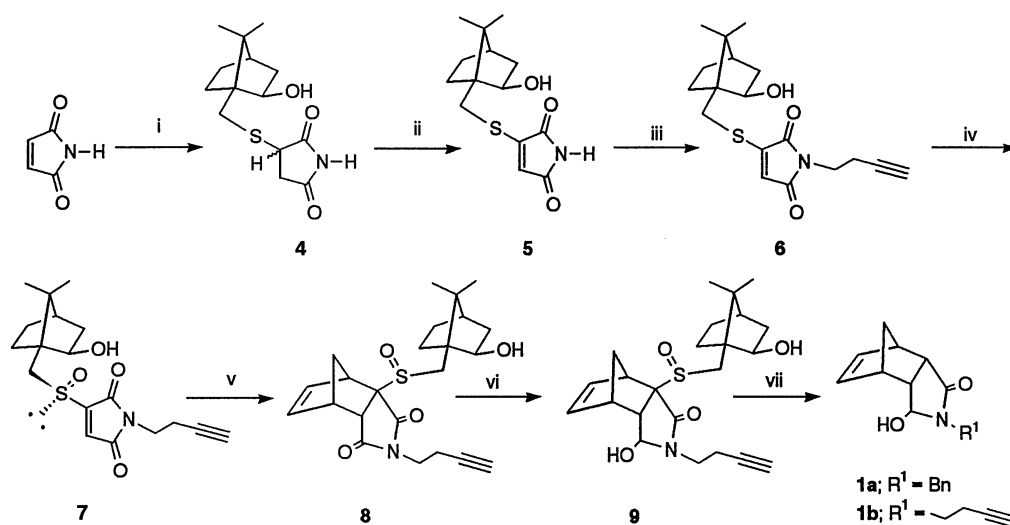
Yoshitsugu ARAI, Tohru KONTANI, and Toru KOIZUMI*

Faculty of Pharmaceutical Sciences, Toyama Medical and Pharmaceutical University,
Sugitani 2630, Toyama 930-01

A new approach to a diastereoselective cationic cyclization by stereocontrol due to the bicyclo[2.2.1]heptene moiety in a chiral γ -hydroxy lactam is described. The diastereoselective reaction followed by Diels-Alder cycloreversion has been successfully applied to a chiral synthesis of (+)-indolizidine and (+)-laburnine (= trachelanthamidine).

Intra-¹⁾ and intermolecular²⁾ nucleophilic additions to acyliminium ions have proven to be a useful method for the synthesis of nitrogen-containing natural products. In particular, a number of syntheses of bicyclic alkaloids such as pyrrolizidine and indolizidine families has been reported;³⁾ however, enantioselective synthesis by the methodology has received much less attention. Very recently there have been a few reports⁴⁾ of asymmetric syntheses based upon the strategy through the nucleophilic addition reaction. However, enantiomeric control in these syntheses has been achieved by the use of the chiral acyliminium ion with a stereogenic center (=chiral auxiliary) appended to the nitrogen atom in the γ -hydroxy lactam. The modest control in these reactions indicates a need for the achievement of high level of diastereoselection. Recently we have shown a short step entry to a chiral γ -hydroxy lactam **1a**.⁵⁾ The lactam **1** obtained has notable structural features: i) nucleophilic addition to the acyliminium generated *in situ* from **1** would take place from the convex face by the steric hindrance due to the fused bicyclo[2.2.1]heptene moiety, and ii) thermal cycloreversion of the bicyclic system could result in the formation of a Δ^3 -pyrrolidinone ring. These structural features enabled us to exploit a new method that effects the addition reaction with a high degree of diastereoselectivity. We describe here a highly diastereoselective acyliminocyclization using a tricyclic lactam **1b** and its application to a chiral synthesis of (+)-indolizidine (**2**)⁶⁾ and (+)-laburnine (**3**).⁷⁾

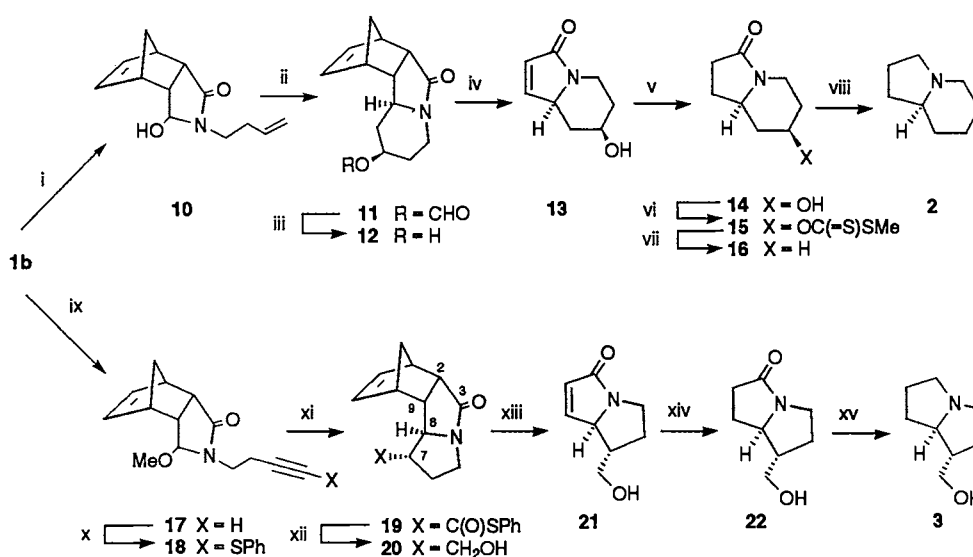
Chiral lactam **1b** was prepared as follows.⁸⁾ Treatment of maleimide with 10-mercaptoisoborneol⁹⁾ gave the succinimide **4** (Scheme 1). Heating of **4** with *N*-chlorosuccinimide (NCS) in carbon tetrachloride resulted in chlorination and spontaneous dehydrochlorination to form the maleimide **5** {mp 182-184 °C, $[\alpha]_D^{25}$ -27.1° (*c* 1, EtOH)}. Imide **5** was coupled with 3-butyn-1-ol under Mitsunobu conditions¹⁰⁾ to afford the *N*-3-butynyl maleimide **6**, which upon exposure to 3-chloroperoxybenzoic acid (MCPBA) gave the sulfoxide **7** {mp 130-131 °C, $[\alpha]_D^{25}$ -59° (*c* 1.05)}.¹¹⁾ Dienophile **7** was allowed to react with cyclopentadiene (ZnCl₂, -75 °C, 0.5 h) giving the adduct **8** {mp 64-66 °C, $[\alpha]_D^{25}$ +3.9° (*c* 2)}, with 96% diastereoisomeric excess. Regioselective reduction of **8** with NaBH₄ (to give **9**) followed by desulphinylation by samarium-induced reduction¹²⁾ afforded the lactam **1b** as a diastereoisomeric mixture.¹³⁾



Scheme 1. i, 10-Mercaptoisoborneol, Et₃N (cat.), CH₂Cl₂, room temp, 6 h (98%); ii, NCS, CCl₄, reflux, 6 h (95%); iii, 3-butyn-1-ol, Ph₃P, tetrahydrofuran (THF), 0 °C, 0.1 h; diethyl azodicarboxylate, THF, 0 °C → room temp, overnight (99%); iv, MCPBA (1.05 mol equiv.), CH₂Cl₂, 0 °C → room temp, 0.5 h (94%); v, cyclopentadiene (5 mol equiv.), ZnCl₂ (1.5 mol equiv.), CH₂Cl₂, -75±5 °C, 0.5 h (93%); vi, NaBH₄ (5 mol equiv.), HCl (cat.), MeOH, 0 °C, 2 h (80%); vii, SmI₂ (5 mol equiv., 0.1 mol dm⁻³ in THF), Bu^tOH (10 mol equiv.), hexamethylphosphoric triamide (10 mol equiv.), room temp, 0.5 h (92%)

The synthesis of **2** is illustrated in Scheme 2. Selective hydrogenation of **1b** over Pd-BaSO₄ afforded the *N*-3-butenyl lactam **10**. The chiral lactam **10** was subjected to an acyliminium cyclization according to the procedure explored by Speckamp,^{1a)} giving the formate **11** as a single product. The spectroscopic data of **10** and **11** were in good agreement with those for racemates.^{1a)} Hydrolysis of **11** produced the alcohol **12**, which upon flash vacuum pyrolysis (FVP, 450 °C / 0.5 Pa) was smoothly transformed into the bicyclic alcohol **13**. The unsaturated amide **13** was hydrogenated to provide **14**, which was then converted into the xanthate **15**. Reduction of **15** with tri-*n*-butyltin hydride gave the amide **16**,¹⁴⁾ which was transformed by reduction to (+)-indolizidine (**2**) { [α]_D²⁴ +9.0° (*c* 0.7, EtOH), lit.^{6a)} [α]_D²³ +9.3 ± 0.6° (*c* 1.77, EtOH)}.

Alternatively, treatment of **1b** with pyridinium *p*-toluenesulfonate¹⁵⁾ and methanol gave the lactam **17** {mp 46-47 °C, [α]_D²⁵ +69.2° (*c* 1.06)} as a single diastereoisomer. The amide **17** was treated with diphenyl disulfide and lithium hexamethyldisilazide to give **18**, which upon exposure to formic acid produced the phenylthioester **19**, exclusively. In contrast to the results of the similar cyclization in monocyclic system,¹⁶⁾ the cyclization proceeded with high diastereoselectivity, which resulted in stereocontrol of the newly formed asymmetric centers, *i.e.* C(7) and C(8) positions in **19**. It seemed likely that the C(7) *sp*² carbon of the enol derivative via the intermediate vinyl cation could be captured by formic acid from the less-hindered convex face. Reduction of **19** with NaBH₄ afforded the alcohol **20**,¹⁷⁾ which was subjected to FVP (500 °C / 1.3 × 10⁻³ Pa) giving the bicyclic amide **21** {mp 62-64 °C, [α]_D²⁶ +38.6° (*c* 0.91)}. Catalytic hydrogenation over platinum oxide (to give **22**) followed by reduction gave (+)-laburnine (**3**) { [α]_D²⁵ +11.1° (*c* 1.1, EtOH), lit.^{7b)} [α]_D +15.4° (*c* 1.44, EtOH), lit.^{7c)} [α]_D²⁰ +13.63° (*c* 1.22, EtOH), lit.^{7d)} [α]_D²² +14.6° (*c* 3.25, EtOH)}, whose ¹H and ¹³C NMR spectra were in good agreement with those of (-)-**3** reported by Ishibashi *et al.*¹⁸⁾



Scheme 2. i, H_2 , Pd-BaSO₄, pyridine (cat.), MeOH, room temp, 15 h (99%); ii, HCO₂H, room temp, 12 h (92%); iii, 2 mol dm⁻³ KOH aq., EtOH, room temp, 2 h (95%); iv, flash vacuum pyrolysis (FVP), 450 °C, 0.5 Pa (83%); v, H_2 , PtO₂ (cat.), MeOH, room temp, 3 h (99%); vi, NaH, imidazole (cat.), THF, reflux, 1 h; CS₂ and then CH₃I, reflux, 0.6 h (80%); vii, Bu₃SnH (1.5 mol equiv.), 2,2'-azobisisobutyronitrile (cat.), benzene, reflux, 10 h (76%); viii, LiAlH₄, Et₂O, reflux, 1 h (74%); ix, pyridinium *p*-toluenesulfonate (cat.), MeOH, room temp, 15 h (90%); x, (PhS)₂, lithium hexamethyldisilazide, THF, -70 °C → room temp, 1 h (96%); xi, HCO₂H, room temp, 15 h (80%); xii, NaBH₄ (2 mol equiv.), MeOH, 0 °C, 0.5 h (94%); xiii, FVP, 500 °C, 1.3 × 10⁻³ Pa (86%); xiv, H_2 , PtO₂ (cat.), EtOH, room temp, 4 h (99%); xv, LiAlH₄ (2 mol equiv.), THF, reflux, 4 h (60%)

In summary, the enantioselective synthesis of bicyclic alkaloids, (+)-2 and (+)-3 has been accomplished by a novel strategy involving acid-catalysed cyclization and Diels-Alder cycloreversion. By choosing an appropriate *N*-substituted maleimide as a chiral dienophile, this synthetic strategy is capable of application to chiral syntheses of a variety of the alkaloids.

We are indebted to Professor J. Kurita, Hokuriku University for his help in the FVP experiment. We thank Professor H. Ishibashi, Kyoto Pharmaceutical University for providing ¹H and ¹³C NMR spectra of (-)-3. This work was partially supported by a Grant-in-Aid for Scientific Research (03670995) from the Ministry of Education, Science and Culture of Japan and by the Yamada Research Foundation, to which we are grateful.

References

- 1) a) B. P. Wijnberg, W. N. Speckamp, and A. R. C. Oostveen, *Tetrahedron*, **38**, 209 (1982); b) W.N. Speckamp and H. Hiemstra, *ibid.*, **41**, 4367 (1985); P.D. Palasz and J.H.P. Utley, *J. Chem. Soc., Perkin Trans. 2*, **1984**, 807; P. Renaud and D. Seebach, *Helv. Chim. Acta*, **69**, 1704 (1986).
- 2) M.Y. Kim, J. E. Starrett, Jr., and S.M. Weinreb, *J. Org. Chem.*, **46**, 5383 (1981); G.A. Kraus and K. Neuenschwander, *J. Chem. Soc., Chem. Commun.*, **1982**, 134; K. Irie, K. Aoe, T. Tanaka, and S. Saito, *ibid.*, **1985**, 633; M. Natsume and M. Ogawa, *Heterocycles*, **20**, 601 (1983); R.M. Williams, P.J. Sinclair,

- D. Zhai, and D. Chen, *J. Am. Chem. Soc.*, **110**, 1547 (1988); K.E. Harding and C.S. Davis, *Tetrahedron Lett.*, **29**, 1891 (1988); K.T. Wanner, A. Kartner, and E. Wadenstorfer, *Heterocycles*, **27**, 2549 (1988); S. Kano, Y. Yuasa, T. Yokomatsu, and S. Shibuya, *J. Org. Chem.*, **53**, 3865 (1988); D.S. Brown, P. Charreau, and S.V. Ley, *Synlett*, **1990**, 749.
- 3) a) Y. Nishimura, "Studies in Natural Products Chemistry," ed by Atta-ur-Rahman, Elsevier, Amsterdam (1989), Vol. 1, p. 227; b) D.F. Taber, R.S. Hoerrner, and M.D. Hagen, *J. Org. Chem.*, **56**, 1287 (1991).
 - 4) R.P. Polniaszek, S.E. Belmont, and R. Alvarez, *J. Org. Chem.*, **55**, 215 (1990); R.P. Polniaszek and S.E. Belmont, *ibid.*, **55**, 4688 (1990); R.P. Polniaszek and S.E. Belmont, *ibid.*, **56**, 4868 (1991).
 - 5) Y. Arai, M. Matsui, T. Koizumi, and M. Shiro, *J. Org. Chem.*, **56**, 1983 (1991).
 - 6) Indolizidine (**2**), although not naturally occurring, has been considered as a simple target molecule in indolizidine alkaloids synthesis: a) B. Ringdahl, A. R. Pinder, W. E. Pereira, Jr., N. J. Oppenheimer, and J. C. Craig, *J. Chem. Soc., Perkin Trans. 1*, **1984**, 1; b) E. Lellman, *Chem. Ber.*, **23**, 2141 (1890); *Ann.*, **259**, 193 (1890); N. J. Leonard and W. J. Middleton, *J. Am. Chem. Soc.*, **74**, 5776 (1952); S. Yamada and T. Kunieda, *Chem. Pharm. Bull.*, **15**, 490 (1967); M. P. Sibi and J. W. Christensen, *Tetrahedron Lett.*, **31**, 5689 (1990).
 - 7) For isolation, see: a) F. Galinovsky, H. Goldberger, and M. Pöhm, *Monatsch. Chem.*, **80**, 550 (1949); b) N. K. Hart and J. A. Lambertson, *Aust. J. Chem.*, **19**, 1259 (1966). For syntheses, see: c) O. Červinka, K. Pelz, and I. Jirkovský, *Collect. Czech. Chem. Commun.*, **26**, 3116 (1961); d) D. J. Robins and S. Sakdarat, *J. Chem. Soc., Perkin Trans. 1*, **1981**, 909.
 - 8) New compounds have been characterized by IR, NMR, and mass spectroscopy, and by microanalyses and/or high resolution mass spectrometry. Optical rotations were taken in chloroform solution unless otherwise stated.
 - 9) E. L. Eliel and W. J. Frazee, *J. Org. Chem.*, **44**, 3598 (1979).
 - 10) O. Mitsunobu, *Synthesis*, **1981**, 1.
 - 11) The oxidation proceeded with high diastereoselectivity ($\approx 100\%$). The absolute stereochemistry of the sulfinyl center in sulfoxide **7** should be tentatively assigned to be *R* (see Ref. 5).
 - 12) G. A. Molander and G. Hahn, *J. Org. Chem.*, **51**, 1135 (1986); Y. Arai, M. Matsui, and T. Koizumi, *J. Chem. Soc., Perkin Trans. 1*, **1990**, 1233.
 - 13) An inconsequential diastereoisomeric mixture of **1b** was separable by chromatography: α -OH isomer, mp 157-158 °C, $[\alpha]_D^{25} +148.8^\circ$ (*c* 2.06); β -OH isomer, mp 143-144 °C, $[\alpha]_D^{24} +68.8^\circ$ (*c* 1.09). The diastereoisomeric ratio was variable depending on the conditions during the reductions and work-up.
 - 14) For a synthesis of (\pm)-**16**, see: N. A. Khatri, H. F. Schmitthenner, J. Shringarpure, and S. M. Weinreb, *J. Am. Chem. Soc.*, **103**, 6387 (1981).
 - 15) M. Miyashita, A. Yoshikoshi, and P. A. Grieco, *J. Org. Chem.*, **38**, 209 (1977).
 - 16) Speckamp reported that the *N*-butenyl pyrrolidinone derivative produced two epimeric cyclized products in a ratio of 4:1 by action of formic acid. P. M. M. Nossin and W. N. Speckamp, *Tetrahedron Lett.*, **1979**, 4411.
 - 17) The optical purity of the alcohol **20** was estimated to be $\approx 100\%$ by the ^{19}F NMR spectrum of its (*S*)-(-)- α -cyano- α -fluorophenylacetic acid ester, see also the determination method: Y. Takeuchi, N. Itoh, H. Note, T. Koizumi, and K. Yamaguchi, *J. Am. Chem. Soc.*, **113**, 6318 (1991).
 - 18) H. Ishibashi, H. Ozeki, and M. Ikeda, *J. Chem. Soc., Chem. Commun.*, **1986**, 654, in Ref. 3a.

(Received September 17, 1991)